

Why PM2.5 Toxicity May Be More Dangerous Than Air Pollution Levels Show

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PM2.5 Toxicity Latest News

- A new study conducted over Kolkata has revealed that the toxicity of PM2.5 air pollutants significantly increases after a certain concentration threshold is crossed, highlighting a sharper health hazard linked to rising pollution levels in Indian cities.

About the Study

- **Title:** *Contrasting features of winter-time PM2.5 pollution and PM2.5-toxicity based on oxidative potential: A long-term (2016–2023) study over Kolkata megacity at eastern Indo-Gangetic Plain*
- **Published in:** *Science of the Total Environment*, December 2024

Significance

- **First-of-its-Kind in India:** This is the first study to assess **how PM2.5 toxicity varies with concentration levels** in an Indian city.
- **Public Health Implications:** The findings call for **urgent pollution control measures**, especially in winter when PM2.5 levels peak.
- **Policy Insight:** Could guide **threshold-based interventions** and health advisories in Indian cities facing high air pollution.

Key Findings of the Study

- **Threshold Level Identified:** The study found that **PM2.5 toxicity sharply increases once the concentration exceeds 70 $\mu\text{g}/\text{m}^3$** .
- **Peak Toxicity Range:** The toxicity continues to rise until the concentration hits 130 $\mu\text{g}/\text{m}^3$, after which it stabilises.
- **No Safe Limit:** Even at concentrations below 70 $\mu\text{g}/\text{m}^3$, PM2.5 poses health risks, though less severe compared to higher levels.
- **Variation Expected:** Thresholds for a sharp rise in toxicity are likely to differ across cities.
 - This is because pollution sources vary — for example, vehicular emissions might dominate in one city, while biomass burning could be a bigger factor in another.

Why Toxicity Rises Sharply Beyond a Threshold

- At lower PM2.5 concentrations, the human body can manage the adverse effects of pollutants more effectively.
- However, beyond a certain threshold ($\sim 70 \mu\text{g}/\text{m}^3$ for Kolkata), the body's natural defence mechanisms become overwhelmed, leading to significantly greater cellular damage, especially in the respiratory system.

Role of Reactive Oxygen Species (ROS)

- **Immune Response:** When pollutants are inhaled, the immune system releases Reactive Oxygen Species (ROS) to combat foreign particles.
- **Double-Edged Sword:** ROS can also harm healthy cells in the body.
- **Antioxidants as Defenders:** To counteract ROS, the body generates antioxidants, which neutralize ROS and protect cells.
- **Oxidative Stress:** When pollutant levels are high, ROS production exceeds the body's antioxidant capacity, leading to oxidative stress, which damages internal cells.

Understanding Oxidative Stress and PM2.5

- The sharp rise in **oxidative stress** at PM2.5 levels beyond 70 $\mu\text{g}/\text{m}^3$ is mainly due to **chemical components from biomass or solid waste burning**.
- **Vehicular emissions** also contribute, but to a **lesser extent** compared to biomass burning.

Need for Oxidative Stress-Based Standards

- While India has set air quality standards for PM2.5 and PM10, there are **no benchmarks for toxicity or oxidative stress**.
- This study aims to establish **toxicity-based threshold values**, helping cities adopt more health-focused pollution control policies.

Current Air Quality Standards: Concentration-Based

- In India, air quality standards for PM2.5 are based only on concentration levels, not on how toxic the air is.

Safe Limits

- Annual average: 40 $\mu\text{g}/\text{m}^3$
- Daily average: 60 $\mu\text{g}/\text{m}^3$

Limitations of Concentration-Based Standards

- Health risks depend not just on how much PM2.5 is present, but also on its chemical composition and toxicity.
- The same concentration may be more harmful in one city than another, depending on the pollutant source (e.g., biomass burning vs. vehicular emissions).

Why Toxicity Matters More

- In Kolkata, PM2.5 levels of 50–60 $\mu\text{g}/\text{m}^3$ may not be significantly more harmful than 30–40 $\mu\text{g}/\text{m}^3$.
- But toxicity spikes sharply after 70 $\mu\text{g}/\text{m}^3$, indicating a real health emergency.

Need for Toxicity-Based Air Quality Standards

- This study strengthens the case for incorporating toxicity thresholds into air quality norms.
- City-specific thresholds should be developed, reflecting local pollutant profiles.
- Such standards could help trigger early warnings or emergency actions when toxicity crosses a critical level, even if concentrations seem “moderate.”

PM2.5 Toxicity FAQs

Q1. What is PM2.5 toxicity?

Ans. It refers to the harmful effects of fine particulate matter based on its chemical composition, not just concentration.

Q2. Why does PM2.5 become more toxic after 70 $\mu\text{g}/\text{m}^3$?

Ans. Because the body’s antioxidant defences are overwhelmed, leading to oxidative stress and increased cellular damage.

Q3. What causes oxidative stress from PM2.5?

Ans. High pollutant levels trigger excessive reactive oxygen species (ROS), which damage cells when antioxidants are insufficient.

Q4. Are India's air quality standards based on toxicity?

Ans. No, current standards are based only on concentration, ignoring the pollutant's chemical toxicity and health impact.

Q5. What policy change does the study suggest?

Ans. It recommends city-specific, toxicity-based air quality thresholds to better assess and respond to health risks.

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